Cytotoxic Glycosides from Albizia julibrissin

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During the course of a study of leguminous plants, cytotoxicity was demonstrated by the crude saponin fraction of *Albizia julibrissin*. Following chromatographic purification, the structures of three novel saponins, julibrosides I–III (**1**–**3**), inclusive of a cytotoxic principle, were elucidated. A comparison of the cytotoxicity of julibrosides (**1**–**3**) and their prosapogenins (**4**–**15**) prepared by alkaline hydrolysis clearly indicated that both an α -L-arabinofuranosyl-(1→4)-[β -D-glucopyranosyl-(1→3)]- α -L-rhamnopyranosyl-(1→2)- β -D-glucopyranosyl ester unit and a monoterpene–quinovopyranosyl moiety are crucial substituents for cytotoxicity among this class of compounds. The hydroxy group at C-16 of aglycon may play an important role in mediating cytotoxicity, and the *N*-acetyl-glucosamine moiety at C-3 seems to enhance activity because **3** showed the strongest cytotoxicity.

The dried stem bark of Albizia julibrissin Durazz. (Leguminosae) (Albiziae Cortex) is used as a tonic in the People's Republic of China and Japan. During our study on the constituents of leguminous plants, we have reported the isolation of six new triterpenoidal saponins, julibrosides A₁₋₄, B₁, and C₁,¹ and three novel acylated triterpenoidal saponins, julibrosides I-III (1-3),² as well as a prosapogenin, prosapogenin 8 (11),² from A. julibrissin. Because a crude saponin fraction obtained from A. julibrissin showed significant (IC₅₀ 12.8 µg/mL) cytotoxicity against the KB cell line, we have determined the cytotoxicity profile of the purified saponins. Furthermore, in order to investigate the relationship between cytotoxicity and compound structure, we have prepared several prosapogenins by alkaline hydrolysis and also tested them for cytotoxicity.



Results and Discussion

The fresh stem bark of *Albizia julibrissin* was extracted with MeOH, and the extract was partitioned

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with *n*-BuOH and H₂O. The aqueous extract was subjected to MCI gel CHP-20P column chromatography to give a crude saponin fraction. In order to further study this class of triterpenoid glycoside, the crude saponin fraction was saponified first using 3% KOH in MeOH. The saponified prosapogenins were separated by column chromatography using MCI gel CHP-20P, Bondapak C₁₈, and Si gel to afford compounds designated as prosapogenins 1-7 (**4**–**10**). Prosapogenins 5 (**8**), 6 (**9**), and 7 (**10**) were identified as julibrosides A₁, A₂, and A₃, respectively, by comparing with each originally isolated sample.¹

Prosapogenin 1 (4) exhibited a $[M + Na]^+$ ion at m/z951 in the positive-ion FABMS. On comparative analysis of the ¹³C-NMR signals of 4 and 9 (Table 1), they were in good agreement, except for the aglycon moiety, which was identified as acacic acid in the case of 4.¹ Therefore, the structure of prosapogenin 1 (4) was determined to be 3-*O*- β -D-xylopyranosyl-(1 \rightarrow 2)- β -D-fucopyranosyl-(1 \rightarrow 6)- β -D-glucopyranosylaccic acid.

Prosapogenin 2 (5) exhibited a $[M + Na]^+$ ion at m/z978 and a $[M + H]^+$ ion at m/z 956 in the positive-ion FABMS. Upon acid hydrolysis with 2 N HCl, **5** gave an acacic acid lactone unit, which was identified with an authentic sample,¹ as well as xylose, arabinose, and *N*-acetyl-glucosamine as component sugars. When the ¹H- and ¹³C-NMR signals of **5** were compared with those of the aglycon portion of **4** (Table 1), they were in good agreement, whereas the sugar signals of **5** were similar to those of **10**, except for the appearance of arabinopyranosyl signals instead of fucopyranosyl signals. Therefore, the structure of **5** was established as $3-O-\beta$ -D-xylopyranosyl-(1 \rightarrow 2)- α -L-arabinopyranosyl-(1 \rightarrow 6)- β -D-2-acetoamido-2-deoxy-glucopyranosylacacic acid.

Prosapogenin 3 (6) exhibited a $[M + Na]^+$ ion at m/z919 in the positive-ion FABMS, suggesting it was the desmethyl derivative of 9. Furthermore, fragment ion peaks at m/z 633 $[M + H - xyl - ara]^+$ and m/z 471 $[M + H - xyl - ara - glc]^+$ were observed. When the ¹³C-NMR signals of 6 (Table 1) were compared with those of 9, the aglycon signals were found to be superimposable. The sugar signals were identical to those of julibroside J₁, which was recently obtained from *A. julibrissin* by Ma *et al.*³ Therefore, the structure of 6

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was determined to be 3-O- β -D-xylopyranosyl- $(1\rightarrow 2)$ - α -L-arabinopyranosyl- $(1\rightarrow 6)$ - β -D-glucopyranosylacacic acid lactone.

Prosapogenin 4 (7) exhibited a $[M + Na]^+$ ion at m/z960 and a $[M + H]^+$ ion at m/z 938 in the positive-ion FABMS. Furthermore, fragment ion peaks at m/z 806 $[M + H - xyl]^+$, 674 $[M + H - xyl - ara]^+$, and m/z468 $[xyl + ara + glcNAc + H]^+$ were observed. When the ¹³C-NMR signals of 7 (Table 1) were compared with those of 5 and 6, resonances due to the aglycon moieties of 6 and 7 were superimposable, whereas the sugar moiety linked at C-3 was found to be identical with that of 5. Therefore, the structure of 7 was determined to be 3-*O*- β -D-xylopyranosyl-(1 \rightarrow 2)- α -L-arabinopyranosyl-(1 \rightarrow 6)- β -D-2-acetoamido-2-deoxy-glucopyranosylacacic acid lactone.

After alkaline hydrolysis of the crude saponin fraction with saturated NaHCO₃ in MeOH, the hydrolysate was separated by successive column chromatography over MCI gel CHP-20P, Bondapak C₁₈, Chromatorex ODS-DU 3050MT, and Si gel to provide prosapogenins 8-12(**11**-**15**), among which the structure of prosapogenin 8 (**11**) has been reported in an ealier paper.²

Prosapogenin 9 (12) exhibited a $[M - H]^-$ ion at m/z1841 in the negative-ion FABMS, and significant fragment peaks at m/z 1709 $[M - H - ara]^-$, 1679 $[M - H - glc]^-$, and 1239 $[M - H - 602]^-$ were observed. By comparing the FABMS of 12 and 2, the molecular ion peak of 12 was 312 mass units less than that of 2. This indicated that 12 lacked the quinovosyl monoterpene unit found in 2. The similar fragment ion peak² at $[M - H - 602]^-$ suggested that the sugar moiety of C-28 was the same as that of 2. When the ¹H- and ¹³C-NMR signals of 12 were compared with those of 2 (Table 2), they were in good agreement, except for lack of a



terminal quinovopyranosyl monoterpene unit affixed to C-21. Inasmuch as signals for the remaining units were in accord with those of **11**, the structure of **12** was determined as 21-O-[(6S)-2-*trans*-2,6-dimethyl-6-O- β -D-quinovopyranosyl-2,7-octadienoyl]-3-O- β -D-xylopyranosyl- $(1 \rightarrow 2)$ - β -D-fucopyranosyl- $(1 \rightarrow 6)$ - β -D-glucopyranosyl- $(1 \rightarrow 3)$ - α -L-arabinofuranosyl- $(1 \rightarrow 4)$ -[β -D-glucopyranosyl- $(1 \rightarrow 3)$]- α -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranosyl ester.

Prosapogenin 10 (13) exhibited a $[M - H]^-$ ion at m/z1857 in the negative-ion FABMS, along with major fragment ion peaks at m/z 1695 $[M - H - glc]^-$ and m/z 1255 [M - H - 602]⁻. By comparison of the FABMS data of 13 and 12, the molecular ion peak of 13 was seen to be 16 mass units larger than that of 12. When the ¹H- and ¹³C-NMR signals of 13 were compared with those of 12 (Table 2), they were in good agreement, except for the tertiary methyl group of a monoterpene acid portion of the molecule. Thus, in the ¹³C-NMR spectrum of **13**, a hydroxymethyl carbon (δ 56.1) was apparent, while a tertiary methyl carbon (δ 12.5) was no longer present. Therefore, the structure of 13 was determined as 21-O-[(6S)-2-trans-2-methyl-6-hydroxymethyl-6-O- β -D-quinovopyranosyl-2,7-octadienoyl]-3-O- β -D-xylopyranosyl-(1 \rightarrow 2)- β -D-fucopyranosyl-(1 \rightarrow 6)- β -Dglucopyranosylacacic acid 28-O-α-L-arabinofuranosyl- $(1 \rightarrow 4)$ -[β -D-glucopyranosyl- $(1 \rightarrow 3)$]- α -L-rhamnopyranosyl- $(1\rightarrow 2)$ - β -D-glucopyranosyl ester.

Prosapogenin 11 (14) exhibited a $[M - H]^-$ ion at m/z1825 in the negative-ion FABMS, suggesting it to be the dehydroxy compound of 12. Furthermore, the fragment ion peaks at m/z 1663 $[M - H - glc]^-$ and 1223 $[M - H - 602]^-$ were 16 mass units less than analogous fragment peaks for 12. When the ¹H- and ¹³C-NMR signals of 14 were compared with those of 12 (Table 2), they were in good agreement, except for those of the aglycon moieties. From the ¹³C-NMR spectrum, the aglycon moiety of 14 was identified to be machaerinic acid, which is a deoxy derivative of acacic acid.¹ Therefore, the structure of 14 was determined as 21-*O*-[(*6S*)-

			Compounds	4-7	,				~
carbon	4	5	6	7	carbon	4	5	6	7
aglycon					sugar				
1	38.7	38.3	38.6	38.6	glc 1	106.6		105.3	
2	26.7	26.2	26.4	26.4	2	75.7		74.9	
3	88.2	88.9	89.3	89.0	3	78.2		76.9	
4	39.5	38.9	39.3	39.2	4	71.2		71.0	
5	55.8	55.5	55.9	55.9	5	76.6		75.7	
6	18.4	18.2	18.2	18.4	6	69.9		68.9	
7	33.4	33.2	32.3	32.5					
8	39.8	39.5	40.3	40.3	glcNAc 1		104.7		104.6
9	47.1	46.8	47.1	47.2	Ž		57.3		57.6
10	36.6	36.6	36.8	36.9	3		74.9		75.2
11	23.8	23.5	23.6	23.7	4		72.3		72.4
12	122.6	122.7	124.8	124.7	5		75.7		76.1
13	144.3	144.4	139.8	140.0	6		69.9		69.3
14	41.9	41.7	43.3	43.3	NHCOCH3		23.2		23.5
15	35.7	35.3	37.5	38.0	NHCOCH3		171.0		171.3
16	74.2	73.8	67.1	66.8					
17	51.6	51.4	50.0	50.0	fuc 1	103.2			
18	40.8	40.6	41.7	41.7	2	82.0			
19	48.3	48.1	43.3	43.0	3	75.1			
20	37.0	36.7	33.9	34.1	4	72.1			
21	73.4	73.5	84.3	83.6	5	71.5			
22	41.7	41.2	26.9	27.1	6	17.4			
23	28.1	27.8	28.6	28.6					
24	15.5	15.2	15.6	15.6	ara 1		101.7	102.2	102.3
25	17.0	16.7	16.1	16.2	2		80.0	79.9	80.3
26	17.1	17.1	16.9	17.0	3		67.0	67.3	67.0
27	27.2	26.9	28.1	28.0	4		72.0	71.9	72.3
28	179.2	179.3	182.5	181.6	5		63.8	64.2	63.8
29	29.9	29.7	28.9	28.9					
30	18.3	18.0	23.4	24.1	xyl 1	106.8	105.7	106.0	106.0
					2	75.7	75.2	74.9	75.4
					3	77.4	77.3	77.2	77.5
					4	70.6	70.4	70.2	70.7
					5	67.0	66.8	66.5	67.0

Table 1. ¹³C-NMR Chemical Shifts for Compounds $4-7^a$

^{*a*} Values are recorded at 100 MHz in pyridine- d_5 .

 $\begin{array}{l} 2\mbox{-}trans\mbox{-}2,6\mbox{-}dimethyl\mbox{-}6\mbox{-}O\mbox{-}\beta\mbox{-}D\mbox{-}quinovopyranosyl\mbox{-}2,7\mbox{-}octadienoyl\mbox{]-}3\mbox{-}O\mbox{-}\beta\mbox{-}D\mbox{-}ylopyranosyl\mbox{-}(1\mbox{-}2)\mbox{-}\beta\mbox{-}D\mbox{-}fucopyranosyl\mbox{-}(1\mbox{-}2)\mbox{-}\beta\mbox{-}D\mbox{-}glucopyranosyl\mbox{-}(1\mbox{-}3)\mbox{-}\rho\mbox{-}L\mbox{-}rabinofuranosyl\mbox{-}(1\mbox{-}4)\mbox{-}[\beta\mbox{-}D\mbox{-}glucopyranosyl\mbox{-}(1\mbox{-}3)\mbox{-}]\mbox{-}\alpha\mbox{-}L\mbox{-}rabinofuranosyl\mbox{-}(1\mbox{-}2)\mbox{-}\beta\mbox{-}D\mbox{-}glucopyranosyl\mbox{-}(1\mbox{-}3)\mbox{-}\\ \alpha\mbox{-}L\mbox{-}rhamnopyranosyl\mbox{-}(1\mbox{-}2)\mbox{-}\beta\mbox{-}D\mbox{-}glucopyranosyl\mbox{-}(1\mbox{-}3)\mbox{-}\\ \alpha\mbox{-}L\mbox{-}rhamnopyranosyl\mbox{-}(1\mbox{-}2)\mbox{-}\beta\mbox{-}D\mbox{-}glucopyranosyl\mbox{-}(1\mbox{-}3)\mbox{-}\\ \end{tabular}$

Prosapogenin 12 (15) exhibited a $[M - H]^-$ ion at m/z1987 in the negative-ion FABMS, thereby indicating that an additional hexose unit was present in this molecule when compared to 14. Furthermore, fragment ion peaks at m/z 1852 [M – H – glc]⁻, 1385 [M – H – 602^{-} , and $1223 [M - H - 602 - glc]^{-}$ were, respectively, 162 mass units larger than those of 14. When the ¹H- and ¹³C-NMR signals of **15** were compared with those of **14** (Table 2), they were in good agreement, except for the sugar moiety at C-3, although the sugar unit linked at C-3 was the same as that of 1. Therefore, the structure of **15** was determined to be 21-O-[(6S)-2*trans*-2,6-dimethyl-6-*O*-β-D-quinovopyranosyl-2,7-octadienoyl]-3-O- β -D-glucopyranosyl-(1 \rightarrow 2)-[β -D-xylopyranosyl- $(1\rightarrow 2)$ - β -D-fucopyranosyl- $(1\rightarrow 6)$]- β -D-glucopyranosylmachaerinic acid 28-O- α -L-arabinofuranosyl- $(1 \rightarrow 4)$ - $[\beta$ -D-glucopyranosyl- $(1\rightarrow 3)$]- α -L-rhamnopyranosyl- $(1\rightarrow 2)$ - β -D-glucopyranosyl ester.

The cytotoxicity of julibrosides I–III (1-3) and their prosapogenins 1–12 (4-15) was examined against the KB cell line, with the IC₅₀ values obtained listed in Table 3. It is apparent that the prosapogenins obtained by alkaline (KOH) hydrolysis (4-10) are completely inactive. On the other hand, some of the acylated triterpene glycosides (3, 11-13) showed significant cytotoxicity. In the case of the genuine saponins (1-3), only julibroside III (3) with a *N*-acetyl-glucosamine unit showed activity. For the prosapogenins obtained by mild alkaline (NaHCO₃) hydrolysis (11-15), the compounds possessing trisaccharide units at C-3 (12-14) were more active than those with tetrasaccharide moieties (11 and 15). The prosapogenins having an acacic acid-type aglycon (11-13) were much more active than those based on machaerinic acid (14 and 15).

A comparison of the cytotoxicity of the julibrosides (1-3) and their prosapogenins (4-15) (Table 3, Figure 1) clearly indicated that both the α -L-arabinofuranosyl- $(1\rightarrow 4)$ - $[\beta$ -D-glucopyranosyl- $(1\rightarrow 3)$]- α -L-rhamnopyranosyl- $(1\rightarrow 2)$ - β -D-glucopyranosyl ester and monoterpene– quinovopyranosyl moieties are crucial substituents required for cytotoxicity. However, the terminal monoterpene–quinovopyranosyl moiety seems not to be necessary for cytotoxicity against KB cells owing to the inactivity of 1 and 2. The hydroxy group at C-16 of the aglycon may play an important role in mediating cytotoxicity among these compounds. Furthermore, the *N*-acetyl-glucosamine moiety at C-3 seems to enhance activity, because 3 showed the strongest cytotoxicity.

Experimental Section

General Experimental Procedures. The optical rotations were measured with a JASCO DIP-360 automatic digital polarimeter. IR spectra were recorded with a JEOL JIR-6500W FT-IR spectrometer. ¹H- and ¹³C-NMR spectra were measured with a JEOL JNM-GX 400 and/or an α -500 FT-NMR spectrometer, and chemical shifts are given on a δ (ppm) scale with Me₄Si as the internal standard. The FABMS were measured with a JEOL DX-300 and/or a JEOL SX102A spectrometer. TLC was performed on precoated Kieselgel 60 F₂₅₄ plates (Merck). GC was performed using a Hewlett

Table 2. ¹³C-NMR Chemical Shifts for Compounds 11–15^a

carbon	11	12	13	14	15	carbon	11	12	13	14	15
aglycon						sugar (C-28)					
C-1	38.7	38.7	38.7	38.8	38.7	glc" 1	95.5	95.4	95.5	95.4	95.3
2	26.6	26.6	26.6	26.7	26.7	$\tilde{2}$	78.9	79.8	78.9	79.0	78.9
3	88.5	88.2	88.2	87.9	88.1	3	76.8	76.9	76.9	76.7	76.9
4	39.5	39.4	39.4	39.6	39.6	4	72.0	72.0	71.6	71.7	72.0
5	55.9	55.9	55.8	55.8	55.8	5	78.2	78.0	78.2	78.3	78.2
6	18.5	18.5	18.5	18.7	18.6	6	62.6	62.3	62.4	62.3	62.6
7	33.4	33.4	33.4	33.4	33.3	rha $(1 \rightarrow 2)$ glc"					
8	39.9	39.9	40.0	39.9	39.8	1	101.7	101.6	101.6	101.6	101.6
9	46.9	47.0	47.0	48.0	47.9	2	70.3	70.4	70.3	70.6	70.6
10	36.9	36.9	36.9	37.0	36.9	3	78.8	78.7	78.7	78.9	78.2
11	23.7	23.7	23.8	23.9	23.8	4	84.2	84.2	84.2	84.6	84.5
12	122.9	122.8	122.9	122.7	122.9	5	69.0	68.9	69.0	68.8	68.8
13	143.2	143.2	143.2	142.9	142.8	6	18.7	18.7	18.7	18.4	18.4
14	41.8	41.8	41.8	42.2	42.1	glc''' (1 \rightarrow 3) rha					
15	35.7	35.7	35.7	28.7	28.4	1	105.5	105.5	105.5	105.6	105.5
16	73.7	73.6	73.7	24.7	24.6	2	75.0	75.0	75.0	75.1	75.0
17	51.5	51.4	51.5	48.6	48.5	3	77.9	77.9	78.0	78.3	78.0
18	40.8	40.8	40.8	41.2	41.1	4	71.1	71.1	71.1	71.2	71.6
19	47.7	47.6	47.7	46.4	46.3	5	78.2	78.0	78.2	78.2	78.0
20	35.1	35.1	35.3	35.4	35.3	6	62.6	62.5	62.6	62.6	71.6
21	76.7	76.5	76.5	76.5	76.5	ara $(1 \rightarrow 4)$ rha					
22	36.2	36.2	36.2	36.3	36.2	1	110.8	110.8	110.8	110.8	110.7
23	27.9	28.0	28.0	28.0	27.8	2	81.8	81.9	81.9	81.9	81.9
24	15.7	15.7	15.7	15.7	15.7	3	78.2	78.0	78.2	78.3	78.2
25	16.7	17.0	17.0	17.1	16.8	4	85.2	85.2	85.2	84.7	84.6
26	17.0	17.2	17.2	17.3	17.0	5	61.8	61.8	61.8	61.9	61.9
27	27.1	27.1	27.1	25.8	25.7						
28	174.3	174.3	174.3	174.8	174.7	MTA ^b 1	167.6	167.6	167.4	167.5	167.4
29	29.0	28.9	29.0	28.7	28.6	2	128.2	128.2	133.5	127.9	127.9
30	19.0	19.0	19.0	18.8	18.7	3	142.3	142.3	145.3	143.0	142.9
sugar (C-3)						4	23.45	23.40	23.40	23.60	23.51
glc 1	104.7	106.5	106.5	106.5	104.7	5	40.5	40.3	40.7	40.3	40.3
2	82.6	75.5	75.5	75.6	82.9	6	79.3	79.3	79.4	79.4	79.3
3	76.8	78.0	78.2	78.3	76.7	7	143.9	143.9	143.8	144.0	143.9
4	71.5	71.5	71.4	71.7	71.6	8	114.7	114.7	114.7	114.8	114.7
5	77.4	77.3	77.3	77.4	77.4	9	12.5	12.5	56.1	12.5	12.4
6	69.6	69.8	69.8	69.8	69.6	10	23.54	23.50	26.60	23.60	23.54
$glc' (1 \rightarrow 2) glc$											
1	105.5				105.6	qui 1	99.1	99.1	99.1	99.2	99.1
2	75.7				75.3	2	75.3	75.3	75.3	75.4	75.3
3	78.0				78.0	3	78.2	78.0	78.2	78.3	78.2
4	71.1				71.2	4	76.7	76.6	76.7	76.7	76.7
5	78.0				78.0	5	72.4	72.4	72.4	72.5	72.4
6	62.3				62.2	6	18.7	18.7	18.7	18.7	18.7
fuc $(1 \rightarrow 6)$ glc											
1	103.2	103.1	103.1	103.3	103.3						
2	82.0	81.9	81.9	82.2	82.3						
3	75.2	75.2	75.2	75.3	75.2						
4	72.4	72.4	72.0	72.1	72.4						
5	71.0	71.0	71.0	71.2	71.1						
6	17.2	17.0	17.0	17.1	17.2						
xyl (1 \rightarrow 2) fuc											
1	106.8	106.6	106.7	106.9	106.9						
2	76.3	75.6	75.6	75.8	75.8						
3	77.8	77.8	77.8	78.0	77.8						
4	70.6	70.5	70.6	70.6	70.6						
5	67.0	67.0	67.0	67.1	67.0						

^{*a*} Values are recorded 100 MHz in pyridine- d_5 . ^{*b*} MTA = Monoterpenoid acid moiety.

Packard HP5890A. The GC conditions were as follows: column, OV-1 (0.5- μ m film bonded, 0.32 \times 30 m); column oven temperature, 230 °C; injection port temperature, 270 °C; detection port temperature, 270 °C; carrier gas, He (2.2 kg/cm²). Column chromatography was carried out on Kieselgel 60 (70–230 mesh and 230–400 mesh), MCI gel CHP-20P (Mitsubishi Chemical Industries, Tokyo, Japan), Bondapak C₁₈ (Waters), Wako gel LP60 C₁₈ (Wako Pure Chemical Industries, Osaka, Japan), Sephadex LH-20 (Pharmacia), and Chromatorex ODS-DU 3050MT (Fuji Silysia, Kasugai, Japan).

Plant Material. The fresh stem bark of *A. julibrissin* Durazz. was collected at the Botanical Garden of

Kumamoto University. Albiziae Cortex, which is the commercially available stem bark of *A. julibrissin*, was purchased from Uchida Wakanyaku Co., Ltd. (Tokyo, Japan).

Cytotoxicity Bioassays Against KB Cells. The tetrazolium-based semiautomated colorimetric assay (MTT assay) developed by Carmichael *et al.*⁴ was used for the in vitro assay of cytotoxicity to KB cells.

Extraction and Isolation. Commercially available Albiziae Cortex (10 kg) was extracted with MeOH twice under reflux. The combined extract was concentrated (537 g) and partitioned with EtOAc and 40% MeOH (380 g). A part of the 40% MeOH extract (180 g) was subjected to column chromatography over MCI gel CHP-

Table 3. Cytotoxicity of Saponins and Prosapogenins from

 Albizia julibrissin

compound	KB cell cytotoxicity [IC ₅₀ , μg/mL (μM)]
julibroside I (1)	>10
julibroside II (2)	>10
julibroside III (3)	1.9 (0.9)
prosapogenin 1 (4)	>10
prosapogenin 2 (5)	>10
prosapogenin 3 (6)	>10
prosapogenin 4 (7)	>10
prosapogenin 5 (8)	>10
prosapogenin 6 (9)	>10
prosapogenin 7 (10)	>10
prosapogenin 8 (11)	9.1 (4.5)
prosapogenin 9 (12)	2.9 (1.6)
prosapogenin 10 (13)	3.7 (2.0)
prosapogenin 11 (14)	3-10 (1.6-5.5)
prosapogenin 12 (15)	>10



Figure 1. Structure–activity relationships for KB cell cytotoxicity of saponins and prosapogenins.

20P using $H_2O\rightarrow 100\%$ MeOH to give fractions 1 to 4. Fraction 3 (32.5 g) was dissolved in 3% KOH–MeOH (50 mL) and refluxed for 10 h. The reaction mixture was carefully neutralized with 1 N HCl and further separated by chromatography over MCI gel CHP-20P ($H_2O\rightarrow 100\%$ MeOH), Bondapak C₁₈ ($H_2O\rightarrow$ MeOH), and Si gel [CHCl₃-MeOH-H₂O (7:3:0.5)] to afford prosapogenins 3 (**6**, 10 mg), 4 (**7**, 11 mg), 5 (**8**, 5 mg), 6 (**9**, 24 mg), and 7 (**10**, 18 mg). Fraction 2 (22.3 g) was further separated by MCI gel CHP-20P ($H_2O\rightarrow$ MeOH) and Bondapak C₁₈ ($H_2O\rightarrow$ MeOH) to afford an acylated saponin fraction (490 mg). The native glycosides were saponified in the same manner as described above to provide prosapogenins 1 (**4**, 10 mg) and 2 (**5**, 9 mg).

Prosapogenin 1 (4): a white amorphous powder; $[\alpha]_D - 8.8^{\circ}$ (*c* 0.50, MeOH); IR ν_{max} (KBr) cm⁻¹ 3465 (OH), 1645 (COOH) cm⁻¹; ¹H NMR (in pyridine- d_s) δ 0.90, 1.03 × 2, 1.35 × 2, 1.44, 1.94 (each 3H, s, *tert*methyl), 1.51 (3H, d, J = 6.6 Hz, fuc – 6), 4.95 (1H, d, J = 7.0 Hz, xyl – 1), 5.03 (1H, d, J = 8.4 Hz, glc – 1), 5.09 (1H, d, J = 6.6 Hz, fuc – 1), 5.65 (1H, br s, H-12); ¹³C-NMR (pyridine- d_s) data, see Table 1; positive-ion FABMS m/z [M + Na]⁺ 951.

Prosapogenin 2 (5): a white amorphous powder; $[\alpha]_{\rm D}$ +1.0° (*c* 0.50, MeOH); IR $\nu_{\rm max}$ (KBr) cm⁻¹ 3470 (OH), 1645 (COOH) cm⁻¹; ¹H NMR (in pyridine- d_5) δ 0.86, 0.99, 1.01, 1.20, 1.37, 1.45, 1.91 (each 3H, s, *tert*methyl), 2.16 (3H, s, NHCO*CH*₃), 3.41 (1H, dd, J = 4.6, 11.5 Hz, H-3), 5.00 (1H, d, J = 7.0 Hz, xyl – 1), 5.05 (1H, d, J = 8.4 Hz, glcNAc – 1), 5.17 (1H, d, J = 5.1Hz, ara – 1), 5.64 (1H, br s, H-12), 8.95 (1H, d, J = 8.8Hz, *NH*COCH₃); ¹³C-NMR (pyridine- d_5) data, see Table 1; positive-ion FABMS m/z [M + Na]⁺ 978, [M + H]⁺ 956, [M + H – xyl]⁺ 824.

Acid Hydrolysis of 5. A small amount of 5 was hydrolyzed by 2 N HCl in a heating bath. After filtration of the mixture, the precipitate was identified as acacic acid lactone by Si gel TLC [CHCl₃–MeOH (10: 1), hexane–acetone (1:1)],¹ while the filtrate was treated with a N₂ stream to give a residue. The residue was converted to the Me₃Si ether of methyl 2-(polyhydroxy-alkyl)-thiazolidine-4-(*R*)-carboxylate and checked by GC,⁵ which showed the sugars to be D-xylose, L-arabinose, and D-*N*-acetyl-glucosamine.

Prosapogenin 3 (6): a white amorphous powder; $[\alpha]_D - 15.5^\circ$ (*c* 0.49, MeOH); IR ν_{max} (KBr) cm⁻¹ 3415 (OH), 1765 (γ -lactone) cm⁻¹; ¹H-NMR (in pyridine- d_5) δ 0.80, 0.83, 1.02, 1.05, 1.09, 1.30, 1.38 (each 3H, s, *tert*methyl), 3.49 (1H, *m*, H-3), 4.90 (1H, d, J = 8.1 Hz, glc – 1), 4.95 (1H, d, J = 7.1 Hz, xyl – 1), 5.05 (1H, d, J = 5.1 Hz, ara – 1), 5.36 (1H, br s, H-12); ¹³C-NMR (pyridine- d_5) data, see Table 1; positive-ion FABMS m/z[M + Na]⁺ 919, [M + H – xyl– ara]⁺ 633, [M + H – xyl– ara – glc]⁺ 471.

Prosapogenin 4 (7): a white amorphous powder; [α]_D -10.1° (*c* 0.47, MeOH); IR ν_{max} (KBr) cm⁻¹ 3455 (OH), 1755 (γ-lactone) cm⁻¹; ¹H-NMR (in pyridine-*d₃*) δ 0.79 × 2, 0.96, 0.99, 1.08, 1.22, 1.39 (each 3H, s, *tert*methyl), 2.24 (3H, s, NHCO*CH*₃), 3.41 (1H, *m*, H-3), 4.95 (1H, d, *J* = 7.3 Hz, xyl - 1), 5.06 (1H, d, *J* = 8.4 Hz, glcNAc - 1), 5.14 (1H, d, *J* = 5.1 Hz, ara - 1), 5.33 (1H, br s, H-12), 9.18 (1H, d, *J* = 8.8 Hz, *NH*COCH₃); ¹³C-NMR (pyridine-*d*₃) data, see Table 1; positive-ion FABMS m/z [M + Na]⁺ 960, [M + H]⁺ 938, [M + H - xyl]⁺ 806, [M + H - xyl - ara]⁺ 674, [xyl + ara + glcNAc + H]⁺ 468.

The fresh stem bark of A. julibrissin (6.3 kg) was extracted with MeOH twice under reflux. The combined extract (379 g) was concentrated and partitioned with *n*-BuOH and H_2O . The aqueous extract (294 g) was subjected to MCI gel CHP-20P column chromatography using $H_2O \rightarrow MeOH$ to give a crude saponin mixture (52 g). A portion (4 g) in saturated NaHCO₃ in MeOH (400 mL) was refluxed for 1 h. The reaction mixture was evaporated to dryness and separated, in turn, by MCI gel CHP-20P ($H_2O \rightarrow MeOH$), Bondapak C_{18} (20%MeOH \rightarrow 70%MeOH), Chromatorex ODS (60%MeOH→70%MeOH), and Si gel column chromatography [CHCl₃-MeOH-H₂O (7:3:0.5 \rightarrow 6:4:1)] to afford prosapogenin 8 (11, 104 mg) as the major component,² along with prosapogenin 9 (12, 34 mg), prosapogenin 10 (13, 68 mg), prosapogenin 11 (14, 29 mg), and prosapogenin 12 (15, 21 mg).

Prosapogenin 9 (12): a white amorphous powder, $[\alpha]_D - 14.6^\circ$ (*c* 0.40, MeOH); ¹H-NMR (in pyridine- d_3) δ 0.99, 1.00, 1.05, 1.08, 1.18, 1.33, 1.56 (each 3H, s, *tert*-methyl), 1.49 (3H, d, J = 6.6 Hz, fuc – 6), 1.60 (3H, d, J = 5.1 Hz, qui – 6), 1.77 (3H, d, J = 6.2 Hz, rha – 6), 1.85 [3H, s, monoterpenoid acid (MTA)-10], 1.91 (3H, s, MTA-9), 4.88 (1H, d, J = 7.7 Hz, qui – 1), 4.90 (1H, d, J = 8.8 Hz, glc – 1), 4.92 (1H, d, J = 8.1 Hz, fuc – 1), 5.00 (1H, d, J = 7.7 Hz, xyl – 1), 5.33 (1H, d, J = 7.7 Hz, glc^{'''} – 1), 5.64 (1H, br s, H-12), 5.92 (1H, br s, rha – 1), 6.06 (1H, d, J = 7.7 Hz, glc^{''} – 1), 6.23 (1H, d, J = 6.6 Hz, ara(f) – 1); ¹³C-NMR (in pyridine- d_5) data, see Table 2; negative-ion FABMS m/z [M – H]⁻ 1841, [M – H – ara(f)]⁻ 1709, [M – H – glc]⁻ 1679, [M – H – 602]⁻ 1239.

Prosapogenin 10 (13): a white amorphous powder; $[\alpha]_D - 13.6^\circ$ (*c* 0.51, MeOH); ¹H-NMR (in pyridine- d_5) δ 0.99, 1.04, 1.05, 1.10, 1.18, 1.33, 1.52 (each 3H, s, *tert*methyl), 1.49 (3H, d, J = 6.2 Hz, fuc – 6), 1.60 (3H, d, J = 5.1 Hz, qui – 6), 1.77 (3H, d, J = 5.5 Hz, rha – 6), 1.91 (3H, s, MTA-10), 4.86 (1H, d, J = 7.7 Hz, qui – 1), 4.92 (1H, d, J = 7.7 Hz, glc – 1), 5.01 (1H, d, J = 8.1Hz, fuc – 1), 5.09 (1H, d, J = 7.0 Hz, xyl – 1), 5.33 (1H, d, J = 7.7 Hz, glc^{'''} – 1), 6.06 (1H, d, J = 8.1 Hz, glc^{''} – 1), 6.20 (1H, d, J = 7.0 Hz, ara(f) – 1); ¹³C-NMR (pyridine- d_5) data, see Table 2; negative-ion FABMS m/z [M – H]⁻ 1857, [M – H – glc]⁻ 1695, [M – H – 602]⁻ 1255, [M – H – 602 – qui – MTA]⁻ 927, [M – H

Prosapogenin 11 (14): a white amorphous powder; $[\alpha]_D - 6.5^\circ$ (*c* 0.47, MeOH); ¹H-NMR (in pyridine- d_3) δ 0.93, 0.97 × 2, 1.09, 1.18, 1.35, 1.41 (each 3H, s, *tert*methyl), 1.49 (3H, d, J = 6.2 Hz, fuc – 6), 1.57 (3H, s, MTA-10), 1.61 (3H, d, J = 5.1 Hz, qui – 6), 1.81 (3H, d, J = 5.5 Hz, rha – 6), 1.86 (3H, s, MTA-9), 4.90 (1H, d, J = 8.1 Hz, qui – 1), 5.04 (1H, d, J = 7.7 Hz, glc – 1), 5.06 (1H, d, J = 6.6 Hz, fuc – 1), 5.07 (1H, d, J = 6.6Hz, xyl – 1), 5.35 (1H, d, J = 7.7 Hz, glc^{'''} – 1), 6.10 (1H, d, J = 7.7 Hz, glc^{''} – 1), 6.24 (1H, d, J = 6.6 Hz, ara(f) – 1); ¹³C-NMR (pyridine- d_3) data, see Table 2 ; negative-ion FABMS m/z [M – H]⁻ 1825, [M – H – glc]⁻ 1663, [M – H – 602]⁻ 1223. **Prosapogenin 12 (15)**: a white amorphous powder; $[\alpha]_D - 22.3^\circ$ (*c* 0.49, MeOH); ¹H-NMR (in pyridine- d_5) δ 0.93, 0.98, 0.99, 1.17, 1.21, 1.31, 1.38 (each 3H, s, *tert*methyl), 1.50 (3H, d, J = 6.2 Hz, fuc – 6), 1.57 (3H, s, MTA-10), 1.60 (3H, d, J = 5.1 Hz, qui – 6), 1.80 (3H, d, J = 5.5 Hz, rha – 6), 1.86 (3H, s, MTA-9), 4.89 (1H, d, J = 7.7 Hz, qui – 1), 5.00 (1H, d, J = 7.7 Hz, glc – 1), 5.02 (1H, d, J = 6.6 Hz, fuc – 1), 5.04 (1H, d, J = 7.0Hz, xyl – 1), 5.34 (1H, d, J = 7.7 Hz, glc^{'''} – 1), 5.40 (1H, d, J = 7.3 Hz, glc' – 1), 6.02 (1H, br s, rha – 1), 6.09 (1H, d, J = 7.7 Hz, glc'' – 1), 6.24 (1H, d, J = 8.1Hz, ara(f) – 1); ¹³C-NMR (pyridine- d_5) data, see Table 2; negative-ion FABMS m/z [M – H]⁻ 1987, [M – H – glc]⁻ 1825, [M – H – 602]⁻ 1385, [M – H – 602 – glc]⁻ 1223.

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